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Received June 7, 2004

The synthesis of new linked bis- and tris-ring tetraazamacrocyclic (bifunctional) reagents for use in an alternative strategy for radiolabelling antibodies is described. For comparison with the above systems, a new single ring bifunctional system incorporating a dioxocyclam ring is also reported.

J. Heterocyclic Chem., **42**, 77 (2005).

Introduction.

Radiolabelled antibodies are of wide interest in cancer research and treatment [1-4] and the use of single ring macrocycles as the isotope-binding components for radioimmunoconjugates has been investigated for some time [5]. Macrocylic systems offer potential advantages over corresponding open-chain systems [6] since both enhanced kinetic and thermodynamic stabilities are characteristic of such metal-bound macrocylic species [7]. Individual macrocylic polyamines, such as the tetraaza derivative, 1,4,8,11-tetraazacyclotetradecane (cyclam) [8], as well as a number of related macrocylic polyaminocarboxylates [9], have been employed to radiolabel various monoclonal antibodies (mAbs) with selected metal isotopes and, recently, a hexaaza cage derivative has also been proposed for such a role [10].

The aim of the investigation now reported was the design and synthesis of new linked 1,4,8,11-tetraazacyclotetradecane (cyclam) derivatives that might yield new radioimmunoconjugates showing enhanced specific activity through binding two or three metal radionuclide ions in both a kinetically and thermodynamically stable manner. In this paper the synthetic procedures for obtaining the new linked macrocylic products are reported; details of their conjugation to mAbs, their interaction with ^{64}Cu and subsequent animal studies employing the resulting radioimmunoconjugates will be reported in due course.

Results and Discussion.

Mono-, bis- and tris-ring macrocylic products, designed for use as bifunctional compounds for the radiolabelling of mAbs have been synthesised. In each of these an anilino group is incorporated in the structure to act as the site for subsequent linking to the mAbs.

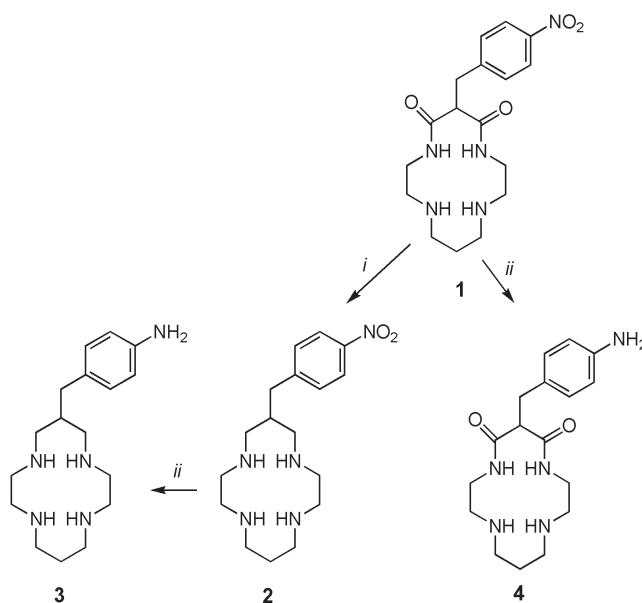
Mono-macrocylic Ring Derivatives.

The single-ring nitro-substituted macrocylic species **1** and **2** were synthesised as described by Moreau *et al.* [11]. The anilino species **3** has been reported previously in the patent literature [12] and its synthesis from **2** (*via* H_2/PdC reduction) described in a recent paper [13]. The

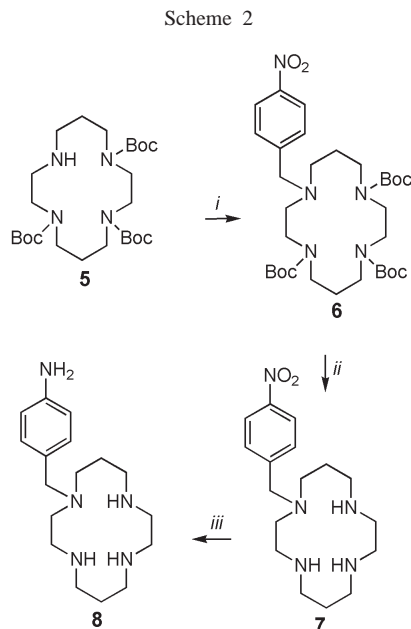
related anilino-substituted, dioxo derivative **4** was generated from **1** in the present study using a similar hydrogenation procedure.

The *N*-substituted single-ring system **8** was prepared using the sequence shown in Scheme 2. Alkylation of the tri-*N*-*boc* protected derivative of cyclam **5** [14] with 4-nitrobenzyl bromide in the presence of caesium carbonate, followed by purification by column chromatography, yielded **6**. The *boc* groups were then cleaved with $\text{HCl}/\text{methanol}$ to afford **7** in 50% overall yield. It needs to be noted that alternative syntheses for **7** using high dilution [15] and boron-protection [14] procedures have been reported previously. Finally, reduction of **7** to yield **8** has also been reported recently [11] involving H_2 (3 atm.) over Pd/C while in the present study sodium borohydride/10% Pd/C was employed.

Scheme 1



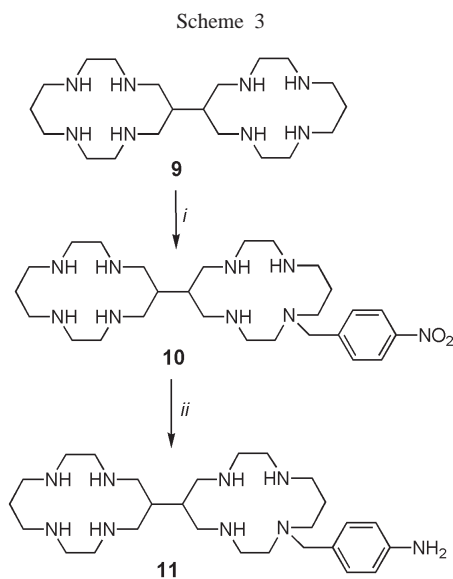
Reagents and Conditions: (i) $\text{BH}_3\text{-THF}$, reflux 20 h. (ii) H_2 at 3 atm, 5% Pd/C .



Boc = *tert*-butoxycarbonyl. Reagents and Conditions (i) Dry acetonitrile, caesium carbonate, BrCH₂C₆H₄NO₂; silica gel-CH₂Cl₂-CH₃OH-25% NH₃OH, 150 : 1 : 0.1. (ii) 3 M HCl-MeOH, NaOH/CHCl₃. (iii) NaBH₄-10% Pd/C, HCl, 10% NaOH-CHCl₃.

Bis-macrocyclic Ring Derivative.

The synthesis of the double-ring macrocyclic ligand system **11**, starting from the previously reported [17,18] bis-ring precursor **9**, was carried out using the sequence shown in Scheme 3. Alkylation of **9** with 4-nitrobenzyl



Reagents and Conditions: (i) BrCH₂C₆H₄NO₂-anhydrous CHCl₃, 10% NaOH-CHCl₃, CH₂Cl₂-di-*tert*-butyl dicarbonate, silica gel-MeOH-CH₂Cl-25% NH₃OH, 200:1:0.1, 10% NaOH-CHCl₃. (ii) NaBH₄-10% Pd/C, HCl, 10% NaOH-CHCl₃.

bromide yielded **10**. As initially isolated, crude **10** was obtained as its hydrobromide salt together with the hydrobromide salts of corresponding bis- and tris-alkylated species as bi-products. Following neutralisation and chloroform extraction of the crude product, attempts at purification using column chromatography on silica gel were unsuccessful due to strong adsorption of the polyamine mixture on the column. In view of this the mixture was reacted with di-*tert*-butyl dicarbonate to convert the secondary amines to their boc-derivatives and hence reduce their polarity. The hepta-'boc' derivative of **10** was successfully separated employing silica gel and the boc groups then removed with methanolic HCl. After workup, pure **10** was obtained as a glassy solid. Hydrogenation (NaBH₄-10% Pd/C) of the product yielded the pendant anilino derivative **11**.

Tris-macrocyclic Ring Derivative.

The three-ring macrocyclic system **15** was synthesised as shown in Scheme 4. The initial step involved the alkylation of the two available amines (4- and 8-positions) of **1** with 4-(chloromethyl)benzoyl chloride to yield **12** which was then reacted with a 2.1-fold molar equivalent of **5** in dry acetonitrile in the presence caesium carbonate to afford **13** in 47% yield after column chromatography on silica gel. The boc groups were then removed (HCl/methanol) to give **14** in 95% yield. Reduction of the nitro substituent of **14** to yield the corresponding anilino derivative **15** was once again achieved by reaction with sodium borohydride in the presence of 10% Pd/C. Workup involved chloroform extraction of the product from 10% sodium hydroxide solution to yield **15** as an oil.

Concluding Remarks.

While a focus of the present study was the development of new linked reagents for the radiolabelling of antibodies for use in cancer therapy, it needs to be noted that linked macrocyclic ligand systems and their metal complexes continue to receive attention in other areas [19]. In particular, very considerable interest in such systems has also been generated by the observation that linked, two-ring, tetraaza macrocycles, and their (selected) metal complexes, inhibit particular HIV strains with low toxicity [20-24]. The present ligand systems are closely structurally related to the bis-ring systems employed in the latter studies and hence may also prove of interest in this context.

EXPERIMENTAL

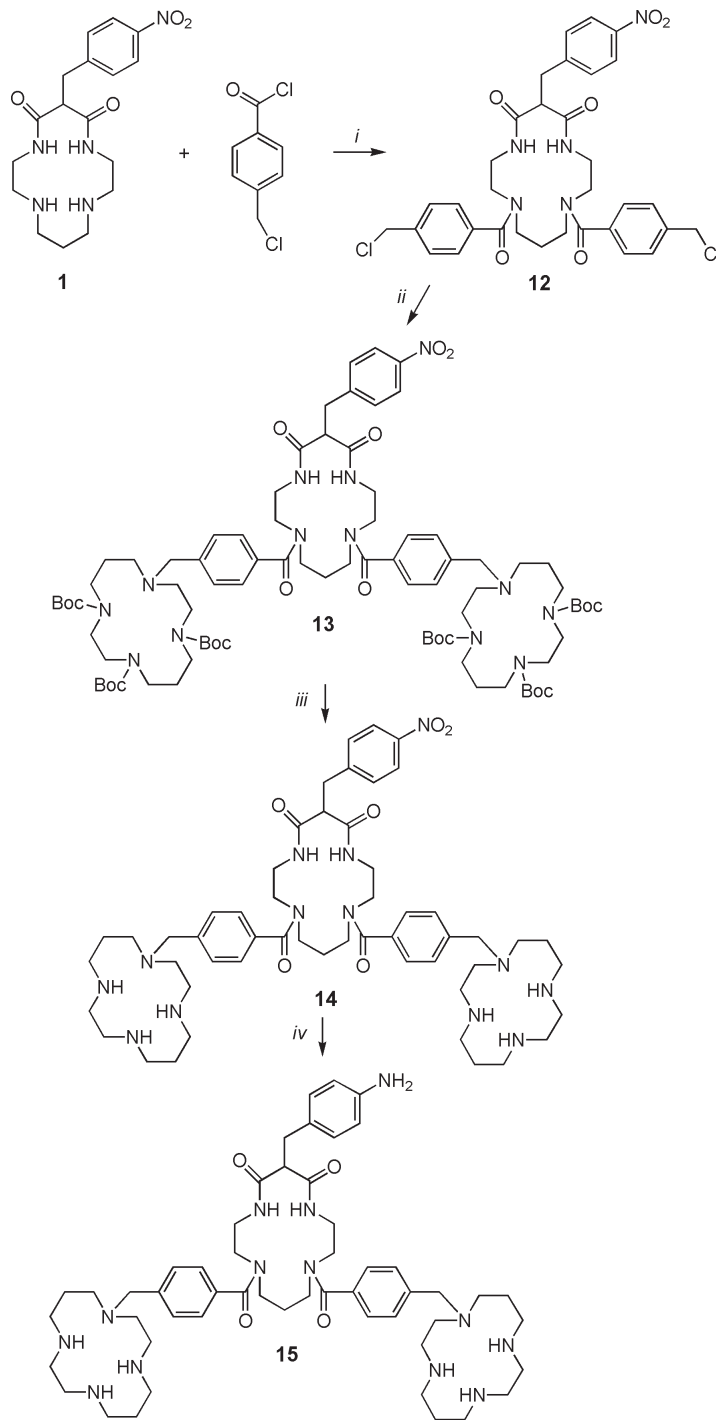
Materials and Apparatus.

Reagents were of the highest grade available commercially. Dry acetonitrile was prepared by shaking acetonitrile with Linde 4A molecular sieves, stirring with calcium hydride until no further hydrogen was evolved, and then fractionally distill-

ing from fresh calcium hydride. Dichloromethane was shaken with portions of sulfuric acid until the acid layer remained colourless; it was then washed with water, aqueous 5% sodium

carbonate, and finally again with water. It was then pre-dried with calcium chloride before being distilled from calcium hydride. Dry tetrahydrofuran was prepared by heating tetrahy-

Scheme 4



Boc = *tert*-butoxycarbonyl. Reagents and Conditions: (i) Dry CH₂Cl₂, triethylamine, 4-(chloromethyl)benzoyl chloride, silica gel, CH₂Cl₂/CH₃OH (50:1). (ii) 5, dry acetonitrile, potassium iodide, caesium carbonate, silica gel-CH₂Cl₂-CH₃OH; 50 : 1; (iii) 3 M HCl-MeOH, NaOH/CHCl₃, (iv) NaBH₄-10% Pd/C, HCl, 10% NaOH-CHCl₃.

dofuran at reflux over phosphorus pentoxide before distillation. The process was then repeated over sodium wire followed finally by fractional distillation. All water used for experimental purposes was of Milli-Q™ grade.

NMR spectra were obtained on Bruker Advance DPX 200, 300 or 400 spectrometers; δ_{H} values are relative to TMS; δ_{C} values are relative to CDCl_3 (at 77.1 ppm), and J values are given in Hertz (Hz). HRMS spectra were obtained on a Crates M25RFA spectrometer and low resolution positive ion (ES) spectra on a Finnegan LCQ-8 spectrometer.

Syntheses.

1,4,8,11-Tetraazaundecane used for the synthesis of **1** was prepared and characterised in a similar manner to that described by Fabbrizzi *et al.* [17]. Cyclam and its precursor 1,5,8,12-tetraazadodecane were synthesised as described by Barefield *et al.* [25]. 6,6'-Bis(1,4,8,11-tetraazacyclotetradecane-5,7-dione), a precursor for **9**, was also prepared as described previously [17].

13-(4-Nitrobenzyl)-1,4,8,11-tetraazacyclotetradecane-12,14-dione (**1**).

This product was prepared as reported previously [11]. Yield 43%. Mass spectrum: m/z 364.3 $[\text{M}+\text{H}^+]$. $^1\text{H-NMR}$ ($\text{CDCl}_3/\text{D}_2\text{O}$): δ 1.62 (2 H, quintet, $\text{NCH}_2\text{CH}_2\text{CH}_2\text{N}$), 2.52-2.79 (8 H, m, CH_2NHCH_2), 3.18-3.54 (7 H, m, NO_2ArCH_2 , COCHCO , CONHCH_2), 7.46 (2 H, J 4.3, d, ArH), 8.13 (2 H, J 4.3, d, NO_2ArH).

Anal. Calcd. for $\text{C}_{17}\text{H}_{25}\text{N}_5\text{O}_4$: C, 56.19; H, 6.93; N, 19.27. Found: C, 55.96; H, 6.80; N, 19.08.

13-(4-Nitrobenzyl)-1,4,8,11-tetraazacyclotetradecane-0.5 CH_3OH -0.5 H_2O (**2**).

This product was prepared as reported previously [11]. Yield 80%. Mass spectrum: m/z 336.5 $[\text{M}+\text{H}^+]$. $^1\text{H-NMR}$ (DMSO): δ 1.64 (2 H, quintet, $\text{NCH}_2\text{CH}_2\text{CH}_2\text{N}$), 2.42-2.76 (19 H, m, $\text{NCH}_2\text{CH}_2\text{CH}_2\text{N}$, $\text{NCH}_2\text{CH}_2\text{N}$, NO_2ArCH_2), 7.49, (2 H, J 4.3, d, ArH), 8.15 (2 H, J 4.3, d, NO_2ArH). $^{13}\text{C-NMR}$ (DMSO): δ 27.7, 30.1, 38.3, 39.6, 48.4, 48.6, 50.3, 55.0, 62.4, 123.6, 129.7, 146.6, 147.8.

Anal. Calcd. for $\text{C}_{17}\text{H}_{29}\text{N}_5\text{O}_4 \cdot 0.5\text{CH}_3\text{OH} \cdot 0.5\text{H}_2\text{O}$: C, 58.31; H, 8.95; N, 19.43. Found: C, 58.58; H, 8.37; N, 19.98.

13-(4-Aminobenzyl)-1,4,8,11-tetraazacyclotetradecane-0.5 CH_3OH (**3**).

This compound was synthesised in a related manner to that reported previously [15]. Yield 92%. Mass spectrum: m/z 306.5 $[\text{M}+\text{H}^+]$. $^1\text{H-NMR}$ (CDCl_3): δ 1.85 (2 H, quintet, $\text{NCH}_2\text{CH}_2\text{CH}_2\text{N}$), 2.44-3.98 (19 H, m, $\text{NCH}_2\text{CH}_2\text{CH}_2\text{N}$, $\text{NCH}_2\text{CH}_2\text{N}$, NH_2ArCH_2), 6.61 (2 H, J 4.1, d, NH_2ArH), 6.93 (2 H, J 4.1, d, ArH). $^{13}\text{C-NMR}$ (CDCl_3): δ 27.6, 37.6, 39.4, 48.1, 50.3, 55.3, 115.2, 129.7, 144.5.

Anal. Calcd. for $\text{C}_{17}\text{H}_{31}\text{N}_5 \cdot 0.5\text{CH}_3\text{OH}$: C, 65.38; H, 10.35; N, 21.78. Found: C, 65.26; H, 10.40; N, 21.34.

13-(4-Aminobenzyl)-1,4,8,11-tetraazacyclotetradecane-12,14-dione- $\text{CH}_3\text{OH} \cdot \text{H}_2\text{O}$ (**4**).

Hydrogenation of **1** to yield **4** as a light yellow solid was carried out as in similar manner to that described for the hydrogenation of **2**. Yield: 89%. Mass spectrum: m/z 334.3 $[\text{M}+\text{H}^+]$. $^1\text{H-NMR}$ (CD_3OD): δ 1.68 (2H, quintet, $\text{NCH}_2\text{CH}_2\text{CH}_2\text{N}$), 2.67-2.80 (8 H, m, CH_2NCH_2), 2.99-3.07 (4 H, m, CONHCH_2), 3.34-

3.58 (3 H, m, NH_2ArCH_2 , COCHCO), 6.76 (2 H, J 4.2, d, NH_2ArH), 7.04 (2 H, J 4.2, d, ArH). $^{13}\text{C-NMR}$ (DMSO): δ 27.4, 32.6, 37.0, 38.2, 47.9, 49.3, 55.2, 113.7, 126.6, 129.0, 146.5, 168.7.

Anal. Calcd. for $\text{C}_{17}\text{H}_{27}\text{N}_5\text{O}_2 \cdot \text{CH}_3\text{OH} \cdot \text{H}_2\text{O}$: C, 56.38; H, 8.67; N, 18.26. Found: C, 56.72; H, 8.23; N, 18.29.

1,4,8-Tris-(*tert*-butoxycarbonyl)-1,4,8,11-tetraazacyclotetradecane (**5**).

This compound was prepared in a similar manner to that described by Guilard *et al.* [14]. Yield: 75%. Mass Spectrum: m/z 501.0 $[\text{M}+\text{H}^+]$. The $^1\text{H-NMR}$ spectrum was identical to that reported previously [14].

11-(4-Nitrobenzyl)-1,4,8-tris-(*tert*-butoxycarbonyl)-1,4,8,11-tetraazacyclotetradecane (**6**).

4-Nitrobenzyl bromide (1.8 g, 8.20 mmol), caesium carbonate (4.8 g, 15.0 mmol) and **5** (3.7 g, 7.40 mmol), were heated in dry acetonitrile (50 mL) at reflux for 4 h under a N_2 stream. The solid was removed by filtration and was washed with warm chloroform. The filtrate was evaporated under reduced pressure and the residue was then partitioned between a mixture of dichloromethane (50 mL) and water (20 mL). The phases were separated and the aqueous solution was shaken with further dichloromethane (3 x 50 mL). The combined dichloromethane extracts were dried (anhydrous Na_2SO_4) and evaporated under reduced pressure to give a light brown oil, which was purified by column chromatography on silica gel (eluting with dichloromethane-methanol-25% ammonia, 150:1:0.1) to give **6** as a light yellow glassy solid. Yield: 3.4 g (74 %). Mass spectrum: m/z 636.4 $[\text{M}+\text{H}^+]$. $^1\text{H-NMR}$ (CDCl_3): δ 1.39 (9 H, s, Bu^t), 1.43 (9 H, s, Bu^t), 1.47 (9 H s, Bu^t), 1.69 (2 H, quintet, $\text{NO}_2\text{ArCH}_2\text{NCH}_2\text{CH}_2\text{CH}_2\text{NBoc}$), 1.72 (2 H, quintet, $\text{Boc-NCH}_2\text{CH}_2\text{CH}_2\text{NBoc}$), 2.40 (2 H, m, $\text{NO}_2\text{Ar-NCH}_2\text{CH}_2\text{CH}_2\text{NBoc}$), 2.63 (2 H, t, $\text{NO}_2\text{ArCH}_2\text{NCH}_2\text{CH}_2\text{N}$), 3.25-3.37 (12 H, m, $\text{NO}_2\text{ArCH}_2\text{NCH}_2\text{CH}_2\text{CH}_2$, CH_2NBoc), 3.63 (2 H, s, NO_2ArCH_2), 7.45 (2 H, J 4.2, d, ArH), 8.16 (2 H, J 4.2, d, NO_2ArH). $^{13}\text{C-NMR}$ (CDCl_3): δ 28.3, 28.4, 46.3, 47.2, 47.9, 52.0, 53.3, 59.2, 123.4, 129.5, 132.4, 147.1, 155.5.

Anal. Calcd. for $\text{C}_{32}\text{H}_{53}\text{N}_5\text{O}_8$: C, 60.45; H, 8.40; N, 11.02. Found: C, 60.54; H, 8.55; N, 11.08.

Anal. Calcd. for $\text{C}_{32}\text{H}_{53}\text{N}_5\text{O}_8$, $[\text{M}+\text{Na}^+]$: 658.3787. Found (FAB): $[\text{M}+\text{Na}^+]$, 658.3795.

11-(4-Nitrobenzyl)-1,4,8,11 tetraazacyclotetradecane-0.25 H_2O (**7**).

A mixture of methanol (18 mL) and concentrated hydrochloric acid (6 mL) was added to **6** (2.0 g, 0.31 mmol). The reaction mixture was allowed to stir overnight at room temperature after which the white solid that formed was collected by filtration. The solid obtained was dispersed in a mixture of 10% aqueous sodium hydroxide (25 mL) and chloroform (50 mL) and the mixture shaken. The phases were separated and the aqueous phase was further shaken with chloroform (6 x 50 mL). The combined chloroform extracts were dried (anhydrous Na_2SO_4) and the solvent removed under reduced pressure. Further purification involved formation of the hydrobromide salt. The title compound was dissolved in absolute ethanol (2 mL) (any undissolved material was removed by filtration). To this solution was added 45% hydrobromic acid in acetic acid (0.5 mL). The mixture was stirred for two h after which the white solid that had formed was collected by filtration. This solid was washed spar-

ingly with cold absolute ethanol and it was then partitioned between a mixture of 10% aqueous sodium hydroxide (10 mL) and chloroform (25 mL). The phases were separated and the aqueous phase shaken with further chloroform (6 x 25 mL). The combined chloroform extracts were dried (anhydrous Na₂SO₄) and evaporated under reduced pressure to give **7** as a light yellow powder. Yield 0.95 g (90%); Mass spectrum: *m/z* 336.7 [M+H⁺]. ¹H-NMR (CDCl₃): δ 1.74 (2 H, quintet, PhCH₂NCH₂CH₂CH₂N), 1.87 (2 H, quintet, NCH₂CH₂CH₂N), 2.51- 2.89 (16 H, m, NCH₂), 3.66 (2 H, s, NO₂PhCH₂), 7.57 (2 H, *J* 4.4, d *ArH*), 8.15 (2 H, *J* 4.3, d, NO₂ArH). ¹³C-NMR (CDCl₃): δ 26.1, 28.3, 47.3, 47.9, 48.6, 49.0, 49.3, 50.9, 53.4, 54.9, 57.7, 123.4, 129.5, 147.1.

Anal. Calcd. for C₁₇H₂₉N₅O₂·0.25H₂O: C, 60.06; H, 8.75; N, 20.60. Found: C, 60.05; H, 8.66; N, 20.36.

11-(4-Aminobenzyl)-1,4,8,11-tetraazacyclotetradecane (**8**).

A solution of sodium borohydride (150 mg) in water (3 mL) was added to a suspension of 10% palladium on charcoal (70 mg) in water (3 mL) under a N₂ stream. The mixture was stirred for five min before a solution of **7** (220 mg, 0.65 mmol) in 2% sodium hydroxide in methanol (6 mL) was added. The reaction was allowed to proceed at room temperature under a N₂ stream for 30 min. The palladium on charcoal was removed by filtration (using a 0.45 μm filter) after which concentrated hydrochloric acid was added dropwise to the filtrate until gas evolution ceased (~ pH 7). This solution was then evaporated under reduced pressure to give a white solid, which was then dispersed in a mixture of 10% sodium hydroxide (20 mL) and chloroform (30 mL). The phases were separated and the aqueous phase was further shaken with chloroform (5 x 30 mL). The combined chloroform extracts were dried (anhydrous Na₂SO₄) and evaporated under reduced pressure to give **8** as a light yellow oil. Yield 180 mg (90%); Mass spectrum: *m/z* 306.5 (MH⁺). ¹H-NMR (CDCl₃): δ 1.82-1.90 (4 H, m, ArCH₂NCH₂CH₂CH₂N, NCH₂CH₂CH₂N), 2.56-2.93 (16 H, m, NCH₂), 3.49 (2 H, s, NO₂ArCH₂), 6.65, (2 H, *J* 4.1, d, NH₂ArH), 7.12 (2H, *J* 4.1, d, ArH). ¹³C-NMR (DMSO): δ 25.9, 28.1, 47.0, 47.8, 48.0, 48.2, 48.6, 49.5, 51.7, 53.6, 56.6, 113.5, 125.3, 129.7, 147.2.

Anal. Calcd. for C₁₇H₃₁N₅; [M+H⁺]: 306.2652. Found (FAB): [M+H⁺], 306.2656.

6,6'-Bis-(1,4,8,11-tetraazacyclotetradecane)-0.25H₂O (**9**).

6,6'-Bis(1,4,8,11-tetraazacyclotetradecane-5,7-dione) was initially prepared in similar manner to that described by Fabbrizzi *et al.* [18]. Yield 24%. Mass spectrum: *m/z* 455.3 [M+H⁺]. ¹H-NMR (CDCl₃): δ 1.64 (4 H, quintet, NCH₂CH₂CH₂N), 2.62 – 3.67 (24 H, m, NCH), 4.02 (2 H, d, CHCO).

Anal. Calcd. for C₂₀H₃₈N₈O₄: C, 52.84; H, 8.43; N, 24.65. Found: C, 52.98; H, 8.11; N, 24.76.

This compound was then converted to **9** using the published procedure [18]. Yield 35.8%; *m/z* 399.5 [M+H⁺]. ¹H-NMR (D₂O): δ 1.66-1.70 (6 H, m, NCH₂CH₂CH₂N, CHCH), 2.55- 2.74 (32 H, m, NCH₂CH₂N, NCH₂CH₂CH₂N).

Anal. Calcd. for C₂₀H₄₆N₈·0.25H₂O: C, 59.59; H, 11.63; N, 27.80. Found: C, 59.34; H, 11.12; N, 27.56.

Compound **9** was also isolated as its octa-hydrochloride salt.

Anal. Calcd. for C₂₀H₄₆N₈·8HCl·2H₂O: C, 33.07; H, 8.05; N, 15.43. Found: C, 33.08; H, 8.37; N, 15.16.

1-(4-Nitrobenzyl)-6,6-bis-(1,4,8,11-tetraazacyclotetradecane) (**10**).

6,6'-Bis-(1,4,8,11-tetraazacyclotetradecane), **9**, (1.0 g, 2.50 mmol) was dissolved in dry chloroform (50 mL). To this solution was added 4-nitrobenzyl bromide (0.54 g, 2.50 mmol) in anhydrous chloroform (50 mL) using motorised syringes employing a 0.62 mL/h flow rate. When the addition was complete, the solution was brought to dryness under reduced pressure after which it was dispersed between a mixture of 10% sodium hydroxide (20 mL) and chloroform (50 mL). The organic phase was separated and the aqueous phase was shaken further with chloroform (6 x 50 mL). The combined chloroform extracts were dried (anhydrous Na₂SO₄) and evaporated to give a light yellow residue, which was then dissolved in dichloromethane (200 mL). To this solution was added di-*tert*-butyl dicarbonate (3.5 g, 17.5 mmol) in dichloromethane. The mixture was stirred for 2 h and again brought to dryness. The intermediate [1-(4-nitrobenzyl)-4,8,11,1',4',8',11'-hepta(*tert*-butoxycarbonate)]-6,6'-bis(tetraazacyclotetradecane) was purified by column chromatography on silica gel (eluting with dichloromethane-methanol-25% ammonium hydroxide, 200:1:0.1). The solvent was removed under reduced pressure and the residue was dissolved in a mixture of methanol (18 mL) and concentrated hydrochloric acid (6 mL). The mixture was stirred overnight and the white solid that formed was collected by filtration and carefully washed with cold methanol. This product was dispersed between a mixture of 10% sodium hydroxide (10 mL) and chloroform (30 mL). The phases were separated and the aqueous phase was shaken with further chloroform (5 x 30 mL). The combined chloroform extracts were dried (anhydrous Na₂SO₄) and the solvent was removed under reduced pressure to give **10** as a light brown glassy solid. Yield: 0.20 g (15%); *m/z* 535 [M+H⁺]. ¹H-NMR (D₂O): δ 1.72-1.90 (6 H, m, NCH₂CH₂CH₂N, CHCH), 2.70- 3.60 (34 H, m, NCH₂CH₂N, NCH₂CH₂CH₂N, NCH₂Ar), 7.54 (2 H, *J* 4.3, d, ArH), 8.31, (2 H, *J* 4.2, d, NO₂ArH). ¹³C-NMR (D₂O): δ 17.4, 25.5, 26.9, 41.1, 46.7, 47.5, 47.6, 47.9, 48.8, 49.4, 52.8, 53.3, 53.5, 54.5, 54.7, 55.5, 58.1, 124.2, 130.7, 147.3, 148.8.

Anal. Calcd. for C₂₇H₅₁N₉O₂, [M+H⁺]: 534.4238. Found (FAB): [M+H⁺], 534.4247.

1-(4-Aminobenzyl)-6,6-bis-(1,4,8,11-tetraazacyclotetradecane) (**11**).

Hydrogenation of **9** to yield **11** was performed in similar manner to that described for the preparation of **8**. (Yield 95%). ¹H-NMR (D₂O): δ 1.72-1.90 (6 H, m, NCH₂CH₂CH₂N, CHCH) 2.70- 3.60 (34 H, m, NCH₂CH₂N, NCH₂CH₂CH₂N, NCH₂Ar), 6.81 (2 H, *J* 5.5, d, ArH), 7.17 (2 H, *J* 5.5, d, ArH). ¹³C-NMR (D₂O): δ 17.4, 21.7, 27.1, 28.8, 40.3, 42.9, 42.7, 48.7, 49.5, 49.7, 50.0, 50.9, 51.5, 55.0, 57.0, 57.5, 59.5, 121.3, 132.4, 133.4, 148.8.

Anal. Calcd. for C₂₇H₅₃N₉, [M+H⁺]: 504.4497. Found (FAB): [M+H⁺], 504.4508.

[13-(4-Nitrobenzyl)-4,8-bis(chloromethyl)benzoyl]-1,4,8,11-tetraazacyclotetradecane-12,14-dione·2H₂O (**12**).

Triethylamine (2.1 g, 20.07 mmol) and a solution of 4-(chloromethyl)benzoyl chloride (3.4 g, 18.25 mmol) in dry dichloromethane were added (over five min) to a suspension of **12** (3.0 g, 8.25 mmol) in dry dichloromethane (200 mL). The mixture was stirred overnight at room temperature to give a clear yellow solution, which was then washed with water (50 mL). The organic phase was further washed with water (50 mL). A white solid formed during the washing procedure, which was collected

by filtration and then dried over phosphorus pentoxide. This crude product was then purified by column chromatography on silica gel (eluting with dichloromethane-methanol, 50:1). The eluant was left to stand and evaporate at room temperature to yield **12** as a white shiny solid, which was again collected by filtration. Yield 3.43 g (64%); Mass spectrum: m/z 668.3 [M+H⁺]. ¹H-NMR (CD₃CN): δ 1.75 (2 H, q, NCH₂CH₂CH₂N), 2.59–3.69 (15 H, m, NCH₂, COCHCO, CH₂ArNO₂), 4.70 (4H, s, ArCH₂Cl), 7.36–7.60 (10 H, d, overlapping, ArH), 8.12 (2H, *J* 4.3, d, NO₂ArH). ¹³C-NMR (CD₃CN): δ 37.9, 38.2, 47.0, 49.5, 50.1, 57.8, 124.8, 128.2, 130.0, 131.4, 138.8, 140.5, 148.1, 148.3, 170.4, 172.6.

Anal. Calcd. for C₃₅H₃₅N₅O₆·2H₂O: C, 56.25; H, 5.58; N, 9.94. Found: C, 56.46; H, 5.61; N, 9.76.

Anal. Calcd. for C₃₅H₃₅N₅O₆: [M+H⁺]: 668.2038. Found (FAB): [M+H⁺], 668.2048.

Compound **13**.

Caesium carbonate (3.2 g, 10 mmol), potassium iodide (0.15 g, 0.90 mmol) and **12** (3.0 g, 4.49 mmol) in dry acetonitrile (10 mL) were heated at reflux under a N₂ stream for 30 min before adding **5** (4.71 g, 9.43 mmol) in a dry acetonitrile (38 mL). The mixture was heated at reflux for 24 h after which the solid was collected by filtration using filter-aid Celite. The solid which collected was washed with warm chloroform. The filtrate was evaporated under reduced pressure and the residue was then partitioned between a mixture of dichloromethane (50 mL) and water (20 mL). The phases were then separated and the aqueous solution was shaken with further dichloromethane (3 x 50 mL). The combined dichloromethane extracts were dried (anhydrous Na₂SO₄) and evaporated under reduced pressure to give a light brown oil which was purified *via* column chromatography on silica gel (eluting with dichloromethane-methanol, 50:1 and 50:2). Yield 3.39 g (47%); Mass spectrum: m/z 1596.9 [M+H⁺]. ¹H-NMR (CDCl₃): δ 1.39 (27 H, s, Bu^t), 1.43 (27 H, s, Bu^t), 1.64 (4H, m, ArCH₂NCH₂CH₂CH₂N), 1.85 (6 H, m, BocNCH₂CH₂CH₂NBoc, CONCH₂CH₂CH₂NCO), 2.33 (4 H, m, ArCH₂NCH₂CH₂CH₂N), 2.59 (4 H, m, ArCH₂NCH₂CH₂NBoc), 2.90–3.52 (43 H, m, BocNCH₂, CONCH₂, CONHCH₂, COCHCO, CHCH₂ArNO₂, ArCH₂N), 7.28–7.43 (10 H, m, overlapping, ArH), 8.09 (2H, *J* 3.9, d, NO₂ArH). ¹³C-NMR (CDCl₃): δ 26.1, 26.7, 28.1, 28.3, 28.4, 28.6, 36.3, 38.4, 45.9, 46.9, 47.5, 48.5, 51.4, 52.7, 57.1, 59.2, 76.6, 79.5, 79.6, 123.7, 126.5, 129.1, 129.6, 134.7, 141.1, 145.6, 146.8, 155.3, 155.5, 169.2, 173.0.

Anal. Calcd. for C₈₃H₁₂₉N₁₃O₁₈: C, 62.42; H, 8.14; N, 11.40. Found: C, 62.35; H, 7.93; N, 11.06.

Anal. Calcd. for C₈₃H₁₂₉N₁₃O₁₈: [M+H⁺]: 1596.9650. Found (FAB): [M+H⁺], 1596.9662.

Compound **14**·CHCl₃.

Conversion of **13** to **14** was performed in an analogous manner to that described for the preparation of **7**. Yield 1.18 g (95 %); Mass spectrum: m/z 996.7 [M+H⁺]. ¹H-NMR (CDCl₃): δ 1.63–1.78 (10 H, m, overlapping, NCH₂CH₂CH₂N), 2.39–2.76 (32 H, m, ArCH₂NCH₂, CH₂NHCH₂), 3.21–3.51 (19 H, m, overlapping, CHCH₂ArNO₂, CONCH₂CH₂CH₂NCO, CONHCH₂CH₂NCO, COCHCO, ArCH₂N), 7.25–7.48 (10 H, d, overlapping, ArH), 8.03 (2 H, *J* 4.0, d, NO₂ArH). ¹³C-NMR (CDCl₃): δ 25.9, 27.7, 29.5, 30.0, 36.0, 37.6, 47.1, 47.6, 48.3, 49.1, 49.3, 51.0, 53.4, 54.2, 55.5, 57.7, 76.6, 123.5, 126.9, 128.9, 129.8, 135.0, 141.1, 146.2, 146.6, 169.2, 172.6.

Anal. Calcd. for C₅₃H₈₁N₁₃O₆·CHCl₃: C, 58.13; H, 7.41; N, 16.32. Found: C, 57.86; H, 7.52; N, 16.69.

Anal. Calcd. for C₅₃H₈₁N₁₃O₆: [M+H⁺]: 996.6506. Found (FAB): [M+H⁺], 996.6494.

Compound **15**.

Hydrogenation of **14** to obtain **15** was achieved in a similar manner to that described for the formation of **8**. (Yield 91%). Mass spectrum: m/z 966.7 [M+H⁺]. ¹H-NMR (D₂O): δ 1.85–2.11 (10 H, m, overlapping, NCH₂CH₂CH₂N), 2.75–3.35 (32 H, m, ArCH₂NCH₂, CH₂NCH₂), 3.52–3.84 (19 H, m, overlapping, CHCH₂ArNO₂, CONCH₂CH₂CH₂NCO, CONHCH₂CH₂NCO, COCHCO, ArCH₂N), 6.79 (2 H, *J* 9.0, d, NH₂ArH), 7.09 (2 H, d, *J* 9.5, ArH), 7.39–7.56 (8 H, d, overlapping, ArH). ¹³C-NMR (DMSO): δ 25.9, 28.0, 30.0, 46.9, 47.5, 48.1, 49.6, 52.4, 54.1, 57.3, 113.7125.9, 126.2, 128.6, 128.9, 135.5, 140.4, 146.7, 169.1, 170.9.

Anal. Calcd. for C₅₃H₈₃N₁₃O₄: [M+H⁺]: 966.6763. Found (FAB): [M+H⁺], 966.6773.

Acknowledgement.

We thank the Australian Institute of Nuclear Science and Engineering and the Australian Research Council for Support.

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